

Purification and Characterization of 50 kDa Extracellular Metalloprotease from *Serratia* sp. ZF03

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Background: Proteolytic enzymes have an important role in variety of physiological and pathological functions. They have been used in therapeutic and pharmaceutical applications. Characterizations of extracellular proteases from various strains of *S. marcescens* indicate that most strains produce a very similar major metalloprotease. This metalloprotease (serrapeptidase, serrapeptase) is an important pharmaceutical agent. Serrapeptase has been used in Asian and European countries for the treatment of inflammatory diseases, cardiovascular disorders, and bacterial infections.

Objectives: In the present study, purification and characterization of extracellular metalloprotease from *Serratia* sp. ZF03 for therapeutic purposes were reported.

Materials and Methods: In this study the protease gene encoding a zinc-metalloprotease was isolated from the previously isolated red-pigmented *Serratia* sp. ZF03. The gene was sequenced and submitted to the GenBank. Proteolytic activity was detected by skim milk agar plate method and zymography. This fragment was found to encode an extracellular zinc-metalloendopeptidase with a molecular weight of approximately 50 kDa. The metalloprotease was purified by ammonium sulfate precipitation and dialysis, and then characterized. The effects of various inhibitors and reagents on protease activity and its kinetic parameters were also determined.

Results: The nucleotide sequence demonstrated that deduced amino acid sequence has a higher identity with those of metalloprotease from serralyisin family. Production of metalloprotease was highest at 48th h of cultivation. Optimum protease activity occurred at a temperature range of 50-55°C and a pH range of 8.0-10. EDTA as a metal chelator, significantly inhibited protease activity. Zymography and inhibition assays showed that metalloprotease is the major secreted protease of *Serratia* sp. ZF03. The kinetic parameters, K_m and V_m , were 0.00105 mg/ml and 0.0531 mM/min, respectively.

Conclusions: Since the metalloprotease of this strain has strong proteolytic properties and good stability, it would be a suitable candidate to be used as an effective drug in the medicine and pharmaceutical industries.

Keywords: Enzyme characterization; Metalloprotease; *Serratia* sp.; Therapeutic applications

1. Background

Inflammatory response, a standard clinical condition, is an important defense mechanism that protects the tissues from invasion by organisms and damaged cells. A variety of enzymes have been used extensively in the management of protein deficiencies and therapeutic applications. Proteolytic enzymes can be administered with non-steroidal anti-inflammatory drugs. Proteolytic enzymes such as chymotrypsin, trypsin and serratiopeptidase play an important role in the inflammatory processes (1). Proteases are found in living organisms including plants,

animals, and microbes. In fact, microbe-derived proteolytic enzymes are extracted from or synthesized by bacterial and fungal sources such as streptokinase, a proteolytic enzyme preparation from β -hemolytic streptococci, brinase from *Aspergillus oryzae*, Sutilain, a proteolytic enzyme isolated from *Bacillus subtilis*, arazyme from *Serratia proteamaculans*, and bacterial toxins like tentoxilysin (tetanus neurotoxin), botulinum neurotoxin and anthrax toxin lethal factor (2-7). Microbial proteases can be divided into different classes, according to activity under acidic, neutral and alkaline environments, chemical characteris-

tic of the catalytic site (serine, threonine, cysteine, aspartic and metallo-peptidases) and mechanism of action. Alkaline proteases are specified as those proteases which are active in a pH range from neutral to alkaline (8,9). Metalloproteases (E.C. 3. 4. 24.) represent the largest class of hydrolases which usually have a catalytic zinc ion in the active site. According to importance and the diversity of metalloproteases, these zinc-dependent enzymes are important targets for biomedical applications (10). *Serratia marcescens* is a Gram-negative bacterium of the family Enterobacteriaceae that secretes a variety of degradative enzymes, such as proteases (serine and thiol proteases, and metalloproteases), lipases, chitinases, hemolysin, and nucleases (11,12). Extracellular zinc metalloendopeptidase, secreted by various *Serratia* strains, specially *S. marcescens*, that has been purified and characterized (11-16). In ten clinical isolates of *S. marcescens*, a 1.5 K bp DNA fragment was amplified, which represented the metalloprotease gene that encoded this protease (11). Serrapeptase (also known as serrapeptidase, serratiopeptidase and *serratia* peptidase) is a metalloprotease that has been purified and characterized from the *Serratia* sp E-15 strain which was isolated from the gut of a larval silkworm (7,12). This enzyme dissolved the rigid proteins of cocoon and allowed the moth to come out (17). It is an effective drug for the treatment of breast engorgement (18) and relieves swelling and pain following maxillary sinus anastomosis (11). Serrapeptase is also used to degrade atherosclerotic plaque and fibrin on the inside of arteries (19,20). In general, this major metalloprotease has been identified in various strains of *Serratia* (11-16) and because of its economic significance, particularly as an anti-inflammatory agent (19), this enzyme is currently being produced commercially and under trade names: Serrapeptase, Aniflazym, Danzen, Serrazyme, Serracel and SerraGold is sold (16,18). Serrapeptase has been permitted to enter as a standard drug in Germany and other European countries for therapeutic consumptions (17). This protease has been used as mucoactive and anti-inflammatory treatment in Japan for 30 years (21). The *Serratia marcescens* metalloprotease is a member of the serralyisin family of proteolytic enzymes, which is a group of metalloproteases with molecular weights of approximately 50 kDa,

and is produced by a variety of species, such as *Erwinia chrysanthemi*, *Yersinia pestis*, *Pseudomonas aeruginosa* and *Yersinia ruckeri*. The serralyisin family belongs to the metzincin clan of metalloproteinases (mainly clan (MA)(M) according to the MEROPS database) (22).

2. Objectives

Because of the therapeutic and commercial significance of metalloproteases, this study attempted to investigate the extracellular metalloprotease produced by the red-pigmented *serratia* sp. ZF03, a strain previously isolated from the hot-springs of high background radiation areas (HBRAs) in Ramsar, Iran (23). The metalloprotease gene was isolated and sequenced, and the extracellular enzyme was subsequently purified and characterized.

3. Materials and Methods

3.1. Bacterial Strain, Cultures, Chemicals and Reagents

The bacterium used, *Serratia* sp. ZF03, was previously isolated from the hot-springs of high background radiation areas in Ramsar, Mazandaran province of Iran (23). Ethylenediaminetetraacetic acid (EDTA), phenylmethylsulfonyl fluoride (PMSF), and iodoacetamide were purchased from Sigma Aldrich (USA). Ammonium sulfate, casein, tryptone, yeast extract, skim milk, glycine, acrylamide, bisacrylamide, ammonium persulfate and all other chemicals were obtained from Merck (Germany). The PCR Master Mix (buffer, Taq DNA polymerase, MgCl₂) was purchased from TaKaRa (Kyoto, Japan).

3.2. Culture Conditions

In order to detect protease production and activity, bacterial culture samples were used to inoculate skim milk agar plates, which were then incubated at 30°C for 24 h. Proteolytic activity was exhibited by a clear zone around or under the colony. A single colony from the *S. marcescens* plate culture was used to inoculate LB broth containing 0.5% (w/v) glucose. Cultivation was carried out at 30°C for 12 h in a rotating shaker, with shaking at 250 rpm. Subsequently, 0.1 ml of the preculture was inoculated into a 1000 ml baffled erlenmeyer flask containing 200 mL of culture

medium, which consisted of the following: tryptone 8 g, yeast extract 4 g, glycine 0.02 g, NaCl 1 g, skim milk 1% (w/v) and 0.5% (w/v) glucose. Skim milk and glucose induced the metalloprotease production (16). Cultures were incubated under the above mentioned conditions for 96 h. Samples of culture supernatant were removed for subsequent analysis at 6 h intervals.

3.3. Amplification of the Metalloprotease Gene by PCR

The genomic DNA of *Serratia* sp. ZF03 was extracted by the HiYield™ Genomic DNA Mini Kit (Real Biotech Corporation, Taiwan), according to manufacturer's instructions forward: 5'-ATG CAA TCT ACT AAA AAG GCA ATT G-3' and reverse: 5'-TTA CAC GAT AAA GTC AGT G-3' primers designed according to the metalloprotease gene of *Serratia* sp. E-15 (X04127.1). Amplification of DNA was carried out using the following PCR program: pre-denaturation at 95°C for 5 min, denaturing at 95°C for 30 s, annealing at 57.8°C for 45 s, extension at 72°C for 55 s (30 cycles) and a final polymerization step at 72°C for 10 min. After amplification, reaction mixture was loaded onto 1% agarose gel. The amplified DNA fragment was then extracted from the agarose gel and sequenced using Sanger's method (Cinagene, Iran). A Sequence homology search was performed using the NCBI BLAST server (<http://www.ncbi.nlm.nih.gov/BLAST>) and ClustalW2-multiple sequence alignment program.

3.4. Protease Purification

For purification of the protease, bacterial culture was centrifuged at 10,000 ×g for 20 min at 4°C. The resulting supernatant was precipitated with ammonium sulfate at different saturation levels (50%, 60%, 67%, 70% and 80%). The most appropriate level of saturation for precipitation of the protease was 67%. Ammonium sulfate (at 67% saturation level) was added slowly to the culture supernatant with stirring at 4°C. After 1 h, the precipitate was collected by centrifugation at 13,000 g for 20 min. The resulting pellet was then dissolved in a minimum of 20 mM Tris buffer (pH 8.0) and dialyzed against the same buffer for 16 h with three buffer changes. Protein concentration was determined by the method of Bradford (24) with bovine serum albumin (BSA) (1mg/ml) as the standard. The protein concentration was esti-

mated by measuring the absorbance at 595 nm using a T90 UV-Visible spectrophotometer (Varian Cary Eclipse, Australia).

3.5. Polyacrylamide Gel Electrophoresis and Zymogram Analysis

SDS-polyacrylamide gel electrophoresis was carried out according to the method of Laemmli with some modifications (25), using a 12% polyacrylamide separating gel and a 5% stacking gel to analyze the protein samples and estimate molecular weight of the protease. Following electrophoresis, the gel was stained with Coomassie Brilliant Blue R-250 and silver nitrate. Detection of protease was also achieved by zymography in the presence and absence of inhibitors such as EDTA, PMSF and iodoacetamide, using gelatin as substrate. Sodium dodecyl sulfate polyacrylamide gels (12%) were co-polymerized with 0.01% (w/v) gelatin. Samples were dissolved in non-reducing SDS-loading buffer without heating. Following electrophoresis, the gels were washed (incubated for 1 h at room temperature on a rotating shaker to remove the SDS and renature the proteinases) with 2.5% Triton-X100 in 50 mM Tris (pH 7.4), 5 mM CaCl₂, and 1 μM ZnCl₂. The buffer was then decanted and the gels were washed by deionized water and incubated for 10 min at 37°C with developing buffer (50 mM Tris (pH 8), 5 mM CaCl₂, 1 αM ZnCl₂ (optional 0.01% sodium azide)), and subsequently stained with Coomassie Brilliant Blue G-250. Zones of proteolysis were detected by destaining.

3.6. Protease Assay

In order to determine proteolytic activity, casein was used at a final concentration of 1.0 % (w/v) in a 0.5 ml assay reaction mixture containing 20 mM Tris-HCl buffer, 10mM CaCl₂, (pH 8.0), and appropriate amount (20 μL) of the enzyme. The reaction mixture was incubated for 10 min at 37°C and terminated by adding 0.5 ml of 10% (w/v) trichloroacetic acid (TCA) on ice. After 20 min, mixture was centrifuged at 18000 ×g for 10 min, and the absorbance of the supernatant was measured at 280 nm.

3.7. Enzyme Characterizations

For determination of optimum temperature, the enzyme samples were incubated at temperatures between 30-65°C for 10 min, using the same stan-

standard assay conditions mentioned above. Thermostability of the enzyme was assessed in 20 mM Tris-HCl, pH 8.0 by incubating the purified protein sample at 40°C and 50°C for 60 min for different time intervals. Enzyme activity was then determined at 37°C as described above. The pH profile of enzyme was determined in different pH values, at 37°C for 10 min. The reaction mixture consisted of 1% (w/v) casein in Britton-Robison universal buffer (0.04 M H₃PO₄, 0.04 M H₃BO₃, 0.04 M CH₃COOH) (26). The activity was determined under standard assay method. Protease enzyme samples were pre-incubated with reagents and inhibitors for 10 min at room temperature, the vials were then transferred to a 37°C water bath and substrate was added. Residual activity was measured under the standard assay conditions. The effect of different metal ions on protease activity was also investigated. Enzyme samples were pre-incubated with different metal ions. Assays were carried out as mentioned above, and the percentage residual activity of the enzyme, in the presence of each metal ion, was subsequently determined.

3.8. Determination of Kinetic Parameters

The enzyme's kinetic parameters, K_m and V_{max} , were determined using different casein concentrations as the substrate under assay conditions. Metalloprotease indicated a Michaelis-Menten type kinetics when hydrolyzing soluble casein. K_m and V_{max} values were obtained by Lineweaver-Burk plot.

4. Results

4.1. Detection of Protease Activity

The production of protease by the *Serratia* sp. ZF03 was investigated. The results of the skim milk agar method showed large clear zones around the colonies after 24 h of incubation (Figure 1). In addition, protease activity was also evaluated in the presence of inhibitors such as PMSF, EDTA, iodoacetamide using gelatin zymography. Proteolytic activity was inhibited completely by EDTA, a known inhibitor of metalloproteases, and partially inhibited in presence of PMSF. Iodoacetamide didn't show any inhibitory effects. Zymogram analysis confirmed that the major secreted protease was metalloprotease. (Figure 2).

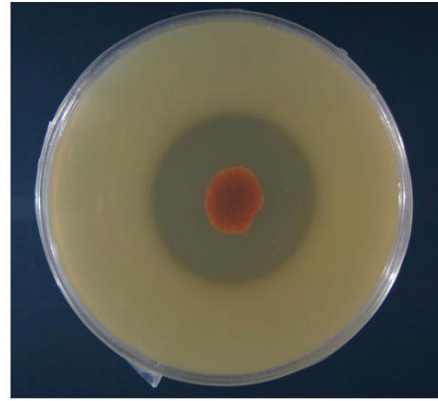


Figure 1. Proteolytic activity as detected by the skim milk agar plate method

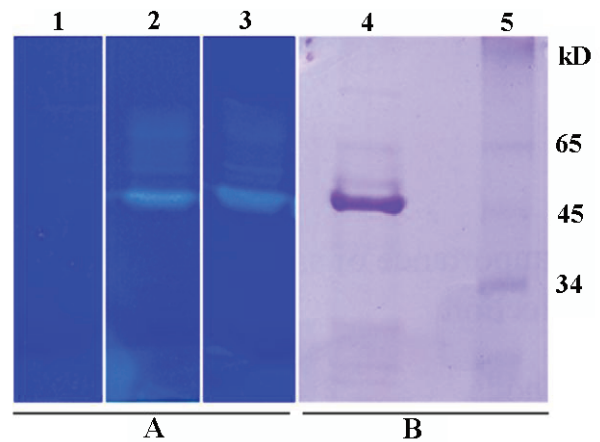


Figure 2. Gelatin-zymography of protease in the presence and absence of inhibitors. Zymogram gel was stained with Coomassie Brilliant Blue G-250. (A) Lane 1, protease activity in the presence of EDTA; Lane 2, PMSF and Lane 3, in the absence of inhibitors. (B) SDS-PAGE; gel stained with Coomassie Brilliant Blue R-250. Lane 4, purified protease and lane 5, molecular weight marker

4.2. Amplification of the Metalloprotease Gene and Sequencing Analysis

Following extraction of the whole genome of *Serratia* sp. ZF03, PCR was performed using primers designed according to the metalloprotease gene of *Serratia* sp. E-15. The amplified DNA fragment was about 1.5 kb in size. The gene was sequenced and submitted to the GenBank, (<http://www.ncbi.nlm.nih.gov/>), under the accession number JX412231. Sequence homology search of the gene showed 97% and 96% maximum identity with that of the metalloprotease

gene of *Serratia marcescens* strain SM6 and *Serratia* (sp. E-15) gene (27). The deduced amino acid sequence of the gene showed 99% identity with the serratiopeptidase from *Serratia* sp. E-15 and *S. marcescens* metalloprotease.

4.3. Time Course of Metalloprotease Production

In order to determine the time course of protease production during bacterial growth, the proteolytic enzyme activity of culture supernatants was measured throughout the growth period, up to the 96th h. The production of total protease was observed at 12th h of growth time, and reached a plateau at the end of the exponential phase. Maximum metalloprotease production was detected at the 48th h of growth (Figure 3).

4.4. Metalloprotease Purification

In order to purify the enzyme, the metalloprotease produced by *S. marcescens* ZF03 was precipitated by ammonium sulfate at the 67% saturation level. The precipitated sample was then dissolved in a minimum of Tris buffer (pH 8.0) and dialyzed. The protease exhibited good purity after this step. The purified enzyme was shown to have a molecular weight of approximately 50 kDa on SDS-PAGE (Figure 4).

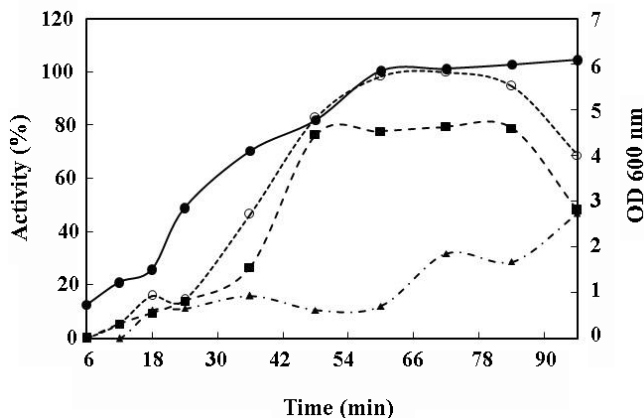


Figure 3. The time course of protease production: Total protease (○), metalloprotease (■), serine protease (▲) and cell growth (●). The protease activity of culture supernatants was measured throughout the growth period, up to the 96th h. In order to determine the type of secreted proteases by the strain, protease activity was also measured in the presence of protease different inhibitors. The growth of *Serratia* sp. ZF03 was also measured by cultivation time at 600 nm

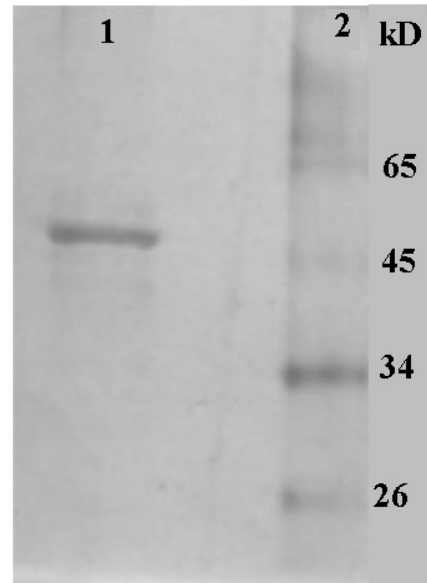


Figure 4. SDS-PAGE analysis of the purified enzyme. Lane 1, purified metalloprotease and lane 2, molecular weight markers. Protein band appeared by silver staining

4.5. Biochemical Characterization

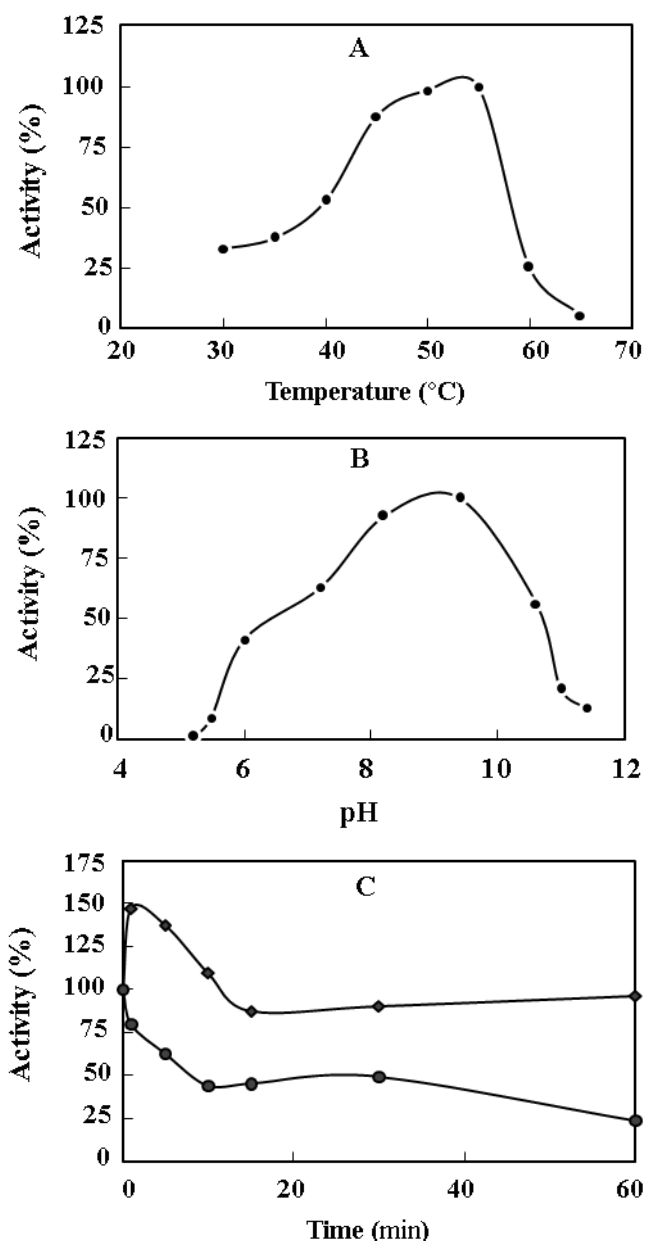
The optimum temperature for metalloprotease activity was observed in the range of 50-55°C (Figure 5A). Optimum metalloprotease activity was also observed at the pH range of 8-10 with a maximum activity at pH 9.0. The effect of metal ions, various reagents and inhibitors on protease activity are also shown in Tables 1 and 2. The use of inhibitors showed that EDTA inhibited significant protease activity compared to PMSF and iodoacetamide (Table 2). Also 1,10 phenanthroline, another inhibitor of metalloprotease, decreased enzyme activity. Iodoacetamide had not only inhibitory effect but also increased the proteolytic activity. Also, iodoacetamide didn't show any inhibition effect on zymogram gel. The activity of the protease was also found to increase in the presence of Co^{2+} , Na^+ and the reagents such as DMSO. The thermostability results showed that the enzyme was stable and active at 40°C, Protease retained approximately 96% of its activity after 60 min at 40°C, whereas 24% of enzyme activity was conserved, when incubated at 50°C after 60 min (Figure 5C). The kinetic parameters, K_m and V_{max} , as determined by the Lineweaver-Burk plot were 0.00105 mg/ml and 0.0531 mM/min, respectively (Figure 6).

Table 1. The effect of metal ions on protease activity. All the metal ions were added as chloride salts (2 mM)

Metal ions	Residual activity (%)
Control	100
Mg ²⁺	68.53
Na ⁺	103.462
K ⁺	97.83
Ca ²⁺	60.72
Al ³⁺	47.05
Ba ²⁺	97.6
Zn ²⁺	32.195
Fe ²⁺	77.07
Fe ³⁺	89.76
Mn ²⁺	85.27
Co ²⁺	108.20

Table 2. The effect of inhibitors and reagents on protease activity

Inhibitors/reagents	Concentration	Residual activity (%)
Control	-	100
SDS	2% (w/v)	53
SDS	5% (w/v)	50
DMSO	2% (v/v)	145
DMSO	5% (v/v)	100
Triton	5% (v/v)	81
Tween80	2% (v/v)	57
Ethanol	2% (v/v)	30
EDTA	5 mM	60
EDTA	10 mM	13
1,10 phenanthroline	2 mM	54
Iodoacetamide	0.5 mM	115
Iodoacetamide	0.1 mM	147
PMSF	2 mM	88
PMSF	1 mM	93

**Figure 5.** Effect of temperature (A) and pH (B) on protease activity. (C) Irreversible thermoinactivation of metalloprotease from *Serratia* sp. ZF03 at 40°C (♦) and 50°C (●)

5. Discussion

Proteolytic enzymes are one of the most prominent groups of industrial enzymes that account for about 60% of the enzyme economics and they are used in variety of industries such as pharmaceutical industry and medicine, detergents, leather and food industry (28). Proteolytic enzymes have been used widely in management of enzyme deficiencies and therapeutic applications. These enzymes

are co-administered with non-steroidal anti-inflammatory agents (2). One of these enzymes is strain-dependent metalloprotease which is secreted by various *S. marcescens* strains (11-16). In this research, the production of protease by the *Serratia* sp. ZF03 (native to Iran) was investigated. First, in order to identify the metalloprotease gene, the DNA fragment (1.5 kbp in size) was isolated from genome of *serratia* according to primers designed based on the metalloprotease

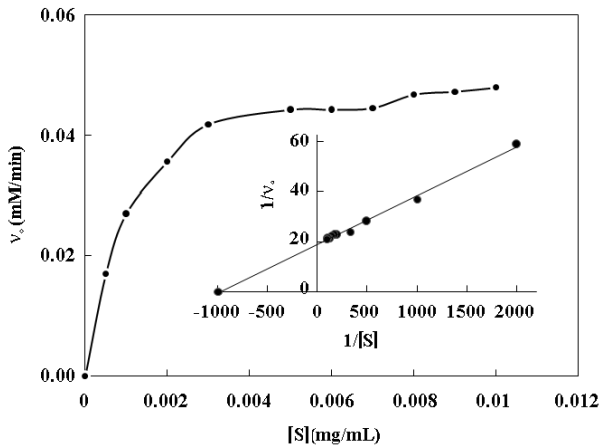


Figure 6. Michaelis-Menten Curve of *Serratia*. sp. ZF03 metalloprotease. K_m and V_{max} values were determined by Lineweaver-Burk plot (inset)

gene of *Serratia* sp. E-15. The molecular weight of the encoded protein, as revealed by SDS-PAGE, was approximately 50 kDa. The results of sequence homology confirmed high identity of this metalloprotease with *S. marcescens* metalloprotease. For determining the type of secreted proteases by the strain, protease activity was also measured in the presence of protease inhibitors such as EDTA, inhibitor of metalloproteases, PMSF (inhibitor of serine-proteases) and iodoacetamide (inhibitor of cysteine proteases). Serine-proteases are inhibited by PMSF at concentrations of 0.1-1 mM. Results showed that total protease activity was observed at 12th h of growth time, had increased exponentially during cell growth and reached to stationary p at 60 h of cultivation time (Figure 3). Because of the low cell growth production the protease production was very low in the first time of cultivation. As total protease, metalloprotease activity increased and remained constant between the 48th-78th h of growth. The protease (total and metallo- protease) production decreased in the stationary phase of cell growth after 84 h, but an increase in activity of serine-protease was observed. This loss of activity might be due to the protease autolysis. The serine-proteases, significantly was not detected throughout the growth period. In fact, maximum metalloprotease production was detected at the 48th h (Figure 3). The results indicate that the most common secreted protease was metalloprotease. Protease activity

in the presence of different inhibitors is comparable to previous reports about *Serratia marcescens* ATCC 25419. Similarly, the major protease identified as metalloprotease, this protease reached high levels at the end of the exponential phase after 18th h of growth. Like some extracellular proteases secreted by *S. marcescens*, this metalloprotease is synthesized during the late exponential phase (16, 29) Bacteria mostly secreted the serine- and metalloproteinases, and to lesser extent, the cysteine proteases (16). In general, the major protease which had been detected in strains of *Serratia* was metalloprotease (11, 16).

Precipitation was an important method for purification of enzymes. One of the effective reagents for precipitation and concentration is ammonium sulfate. Proteins can be isolated with different concentrations of ammonium sulfate. In purification of protease from *Serratia* sp. ZF03, bacterial cultures were removed at the 48th h of growth. After centrifugation of cultures, different saturation levels (50%, 60%, 67%, 70% and 80%) of ammonium sulfate were used for enzyme precipitation.

Enzymatic activity and SDS-PAGE analysis of precipitated supernatants confirmed that among those, 67% ammonium sulfate was the best. The result of SDS-PAGE showed that 50 kDa metalloprotease of this fraction (67% ammonium sulfate) was found to be major band. Similar report presented about purification of a 50 308 Da alkaline protease produced by *S. marcescens* S3-R1 from the Korean ginseng rhizosphere (30).

Metalloprotease from *Serratia*. sp. ZF03 was shown to be optimally active in the range of 50-55°C (Figure 5A). At 45°C retained 85% of its optimal activity. Protease activity declined abruptly to about 25% of the maximal value at 60°C. The optimal temperatures of most bacterial metalloproteases are in the range of 45-60°C. The metalloprotease activity at pH ranging from 5.0 to 9.0 increased steadily until achieving the maximum activity at approximately pH 9.0. This metalloprotease was optimally active at the pH range of 8.0-10.0 (with a maximum activity at pH 9.0) (Figure 5B) similar to most alkaline metalloproteases (7, 12, 16, 31-35). For example, Metalloprotease produced by *S. marcescens* ATCC 25419 has pH profile with the maximum activity at pH 8.5 and temperature profile (optimal activity at 45°C) (16). Also, similar results were obtained from *S.*

marcescens MH6 but the optimum temperature for metalloprotease was the range of 40-45°C and optimal pH was 8.0 (15). Other reports revealed metalloprotease from *S. marcescens* S3-R1 was optimally active at 40°C and pH 8.5 (30). Comparison of results show that our metalloprotease had a higher optimal pH and temperature.

The use of inhibitors (Table 2) showed that EDTA inhibited significantly protease activity compared to PMSF and iodoacetamide. Also, 1, 10 phenanthroline, another inhibitor of metalloprotease, decreased enzyme activity. Iodoacetamide had not only inhibitory effect but also increased the proteolytic activity. Also, iodoacetamide didn't show any inhibition effect on zymogram gel. Considering that iodoacetamide is inhibitor of cysteine proteinases, This result suggests that no cysteine-protease was detected among the secreted enzymes in this strain. This data demonstrated that the metalloproteinases are the major proteases secreted by *Serratia* sp. ZF03. Moreover, the results of zymogram analysis strongly showed the presence of metalloprotease in this strain.

In this study, the effect of metal ions and reagents on protease activity showed that Co^{2+} , Na^+ marginally and reagents such as Iodoacetamide 0.1 mM and 2% DMSO significantly increased metalloprotease activity. Enzyme behavior in the presence of metal ions and reagents is comparable to previous reports. It was shown that the activity of alkaline proteinases of *P. aeruginosa* and metalloprotease of *Xenorhabdus* were increased by Co^{2+} (36, 37). Therefore, maybe Co^{2+} is able to reactivate the activity of protease after inhibition. Similar patterns of activation and stability by DMSO were observed for proteases from solvent-tolerant strains of *S. marcescens* MH6 and *P. aeruginosa* (15, 37). In the presence of ethanol, SDS detergent and Zn^{2+} , the activity of metalloprotease decreased to 30%, 50% and 30.19% of its initial activity, whereas the addition of Ba^{2+} and K^+ ions and 1mM PMSF had no considerable effect on protease activity. Fe^{2+} , Fe^{3+} and Mn^{2+} ions and 2 mM PMSF partially inhibited the activity of protease. Protease activity was inhibited by Zn^{2+} , like that of metalloprotease reported from *Serratia marcescens* MH6. In contrast, The activity of the protease from strain ATCC 25419 increased in the presence of Zn^{2+} (16). This activation/inactivation

by Zn^{2+} may be due to special structure in the catalytic site. 1% SDS detergent inhibited strongly the activity of MH6 protease whereas relative activity of our metalloprotease in presence of 5% SDS retained 50% of initial activity which suggested that cationic residues of the protease from *S. marcescens* MH6 was more than that of protease produced by ZF03 strain (15). Reagents like DMSO solvent and metal ions such as Co^{2+} , Na^+ , Ba^{2+} , Mn^{2+} , K^+ may protect the protease conformation against environmental changes. Therefore protease can be stable and active at high temperatures in the presence of reagents. In thermoinactivation process, protease activity increased abruptly after 1 min of incubation at 40°C then it lost activity, it reached 98% of its initial activity after 10 min of incubation. This activation was probably due to the induction of flexibility in its relatively rigid structure of protease. Protease retained approximately 90% and 96% of its activity after 30 and 60 min respectively at 40°C, whereas 24% of enzyme activity was conserved, when incubated at 50°C after 60 min (Figure 5C). Such thermal stability was observed about metalloprotease of *S. marcescens* S3-R1 at 40°C (30). Considering stability of protease at 40°C, protease can be resistant and active in physiological and biochemical conditions in the human body.

In conclusion, the above results show that the protease investigated in this study, is an alkaline e. metalloproteas. Previous studies have shown that metalloproteases isolated from *Serratia* strains have anti-inflammatory effects and thus therapeutic properties (12, 19). Proteolytic enzymes that catalyze the hydrolysis of peptide bonds are present in all organisms and are involved in a large variety of physiological and biochemical functions. Most therapeutic enzymes are extracted from animals and microorganisms (10). Consequently, with respect to the high level of metalloprotease being produced by the *Serratia* sp. ZF03 strain, and its strong proteolytic properties, the metalloprotease identified in this study can be a potential candidate for use in medicine and the pharmaceutical industries. Due to the important role of this metalloprotease, further enzymatic studies, structural analysis, formulation and immobilization of enzyme can be carried out in order to understand the mechanism of its activity and industrial applications.

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Authors' Contribution

Salari N, performed her thesis. Dr. Hasannia supervised in this research. Dr. Sajedi and Dr. Akbari were thesis advisors.

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